

APPENDIX

Appendix 1. List of chemicals and materials used in the study.

The chemicals used were analytical grade unless specified.

Chemicals/materials	Source
Absolute Ethanol	Wako pure chemical industries Ltd., Japan
Azoxymethane	Sigma Chemical Co., U.S.A.
Barium chloride	M&B Ltd., England
Bis-Tris [bis(2-hydroxymethyl)imino-tris(hydroxymethyl)methane]	Sigma Chemical Co., U.S.A.
Bovine serum albumin	Sigma Chemical Co., U.S.A.
<i>tert</i> -butylhydroperoxide	Sigma Chemical Co., U.S.A.
1-Chloro-2,4-dinitrobenzene	Wako pure chemical industries Ltd., Japan
Chloroform	Wako pure chemical industries Ltd., Japan
Cupric sulfate	Nakarai Chemicals Ltd., Japan
2,6-Dichlorophenolindiphenol	Sigma Chemical Co., U.S.A.
Dicoumarol	Sigma Chemical Co., U.S.A.
Diethyl ether	J.T. Baker, U.S.A.
Dimethyl sulfoxide	E. Merck, Germany
Dipotassium hydrogen phosphate	Wako pure chemical industries Ltd., Japan
Disodium hydrogen phosphate anhydrous	Wako pure chemical industries Ltd., Japan
Ethylene diamine-N,N,N',N'-tetra acetic acid	Wako pure chemical industries Ltd., Japan
Flavin adenine dinucleotide	Sigma Chemical Co., U.S.A.
Folin-phenol reagent	Wako pure chemical industries Ltd., Japan

Glucose-6-phosphate	Sigma Chemical Co., U.S.A.
Glucose-6-phosphate dehydrogenase	Sigma Chemical Co., U.S.A.
Glutathione	Sigma Chemical Co., U.S.A.
Glycine (Amino acetic acid)	Wako pure chemical industries Ltd., Japan
Hydrochloric acid	Wako pure chemical industries Ltd., Japan
Isoamylalcohol	Wako pure chemical industries Ltd., Japan
Magnesium chloride	E. Merck, Germany
Methylene blue	Wako pure chemical industries Ltd., Japan
N ⁷ -methylguanine	Sigma Chemical Co., U.S.A.
Nicotinamide adenine dinucleotide reduced form	Sigma Chemical Co., U.S.A.
<i>p</i> -Nitrophenol	Sigma Chemical Co., U.S.A.
<i>p</i> -Nitrophenyl- β -D-glucuronide	Sigma Chemical Co., U.S.A.
O ⁶ -methylguanine	Dr. K. Ishizaki and Dr. M. Ikenaka, Kyoto University, Japan
Phenol crystal	Wako pure chemical industries Ltd., Japan
Potassium chloride	J.T. Baker, U.S.A.
Potassium dihydrogen Phosphate	Wako pure chemical industries Ltd., Japan
Potassium sodium tartrate	Wako pure chemical industries Ltd., Japan
Proteinase K	Worthington biochemicals Co., U.S.A.
Ribonuclease A	Worthington biochemicals Co., U.S.A.
Ribonuclease T ₁	Worthington biochemicals Co., U.S.A.
Sodium cacodylate	Wako pure chemical industries Ltd., Japan
Sodium chloride	Wako pure chemical industries Ltd., Japan
Sodium citrate	Wako pure chemical industries Ltd., Japan
Sodium dithionite	Wako pure chemical industries Ltd., Japan
Sodium dodacyl sulfate	Wako pure chemical industries Ltd., Japan
Sodium hydroxide	Wako pure chemical industries Ltd., Japan
2-Thiobarbituric acid	Wako pure chemical industries Ltd., Japan

Trichloroacetic acid	Wako pure chemical industries Ltd., Japan
Tris(hydroxymethane) aminomethane	Sigma Chemical Co., U.S.A.
UDP-glucuronic acid	Sigma Chemical Co., U.S.A.

Appendix 2. List of instrument used in the study

Instrument	Model	Source
Autoclave	SS-240	Tomy Sciko Co. Ltd., Tokyo, Japan
Fluorescence detector	RF-530	Shimadzu, Japan
High Performance Liquid Chromatography	LC-5A	Shimadzu, Japan
Light microscope		Olympus, Japan
pH meter	701A	Orioh Inc., U.S.A.
Refrigerator	GR-2000TG	Toshiba, Thailand
Refrigerator (-80°C)		Sanyo, Thailand
Spectrophotometer		Shimadzu, Japan
Super speed centrifuge	RC2-B	Ivan Sorval Inc., U.S.A.
water bath	Type 1 No. 7095	Yazama, Japan

Appendix 3 Reagent preparation

10% Formaldehyde in PBS (2 liters)

PBS	1800	ml
formaldehyde	200	ml

Phosphate buffer saline (PBS) pH 7.4 (1 liter)

NaCl	8	g
KCl	0.2	g
Na ₂ HPO ₄	1.44	g
KH ₂ PO ₄	0.24	g
dH ₂ O	800	ml

adjust pH 7.4 using 1 N HCl then adjust volume to 1 liter

Methylene blue dye 0.2% in saline

Methylene blue	1.0	g
Saline	500	ml

0.1 M Tris-HCl buffer pH 8.0

Tris	12.11	g
dH ₂ O	800	ml

adjust pH 8.0 using 6N HCl and then adjust volume to 1 liters.

10 mM Sodium cacodylate pH 7.0

Sodium cacodylate	0.214	g
dH ₂ O	80	ml

adjust pH 7.0 and then adjust volume to 100 ml

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Isoamylalcohol (3-methyl-1-butanol)	1	ml
Chloroform	24	ml

0.05 M EDTA

EDTA.2Na	18.6	g
dH ₂ O	1000	ml

50 mM Bis-Tris-1 mM MgCl₂ pH 6.5

Bis-Tris (50 mM)	1.046	g
MgCl ₂ .6H ₂ O (1 mM)	0.020	g
dH ₂ O	80	ml

adjust pH 6.5 and then adjust volume to 100 ml

10% Sodium dodesyl sulfate-10mM EDTA-20mM Tris-HCl pH 7.4

Tris (20 mM)	1.211	g
EDTA (10 mM)	1.861	g
Sodium dodesyl sulfate	5.0	g
dH ₂ O	400	ml

adjust pH 7.4 with HCl and then adjust volume to 500 ml

1/100 SSC-0.1 mM EDTA

Sodium chloride (0.15 M)	0.88	g
Sodium citrate (0.015 M)	0.44	g
EDTA (0.1 mM)	0.0037	g
dH ₂ O	100	ml

4 mM Ammonium Formate pH 3.0

HCOONH ₄	0.252	g
dH ₂ O	800	ml

adjust pH 3.0 with HCl and then adjust volume to 1000 ml

6 mM tert-Butylhydroperoxide (t-BHP)

t-BHP	7.72	μl
PBS pH 7.4	10	ml

0.1 M Potassium phosphate buffer pH 7.8

K_2HPO_4	1.742 g in H_2O 100 ml
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KH_2PO_4	1.361 g in H_2O 100 ml
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adjust pH 7.8 using KH_2PO_4

8.0% Sodium dodesyl sulfate (SDS)

SDS	8 g
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d H_2O	100 ml
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20% Acetic acid

Acetic acid	20 ml
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d H_2O	80 ml
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0.8% Thiobarbituric acid (TBA)

Thiobarbituric acid	0.8 g
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d H_2O	95 ml
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6N NaOH	5 ml
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Appendix 4 Morphology of Aberrant crypt foci

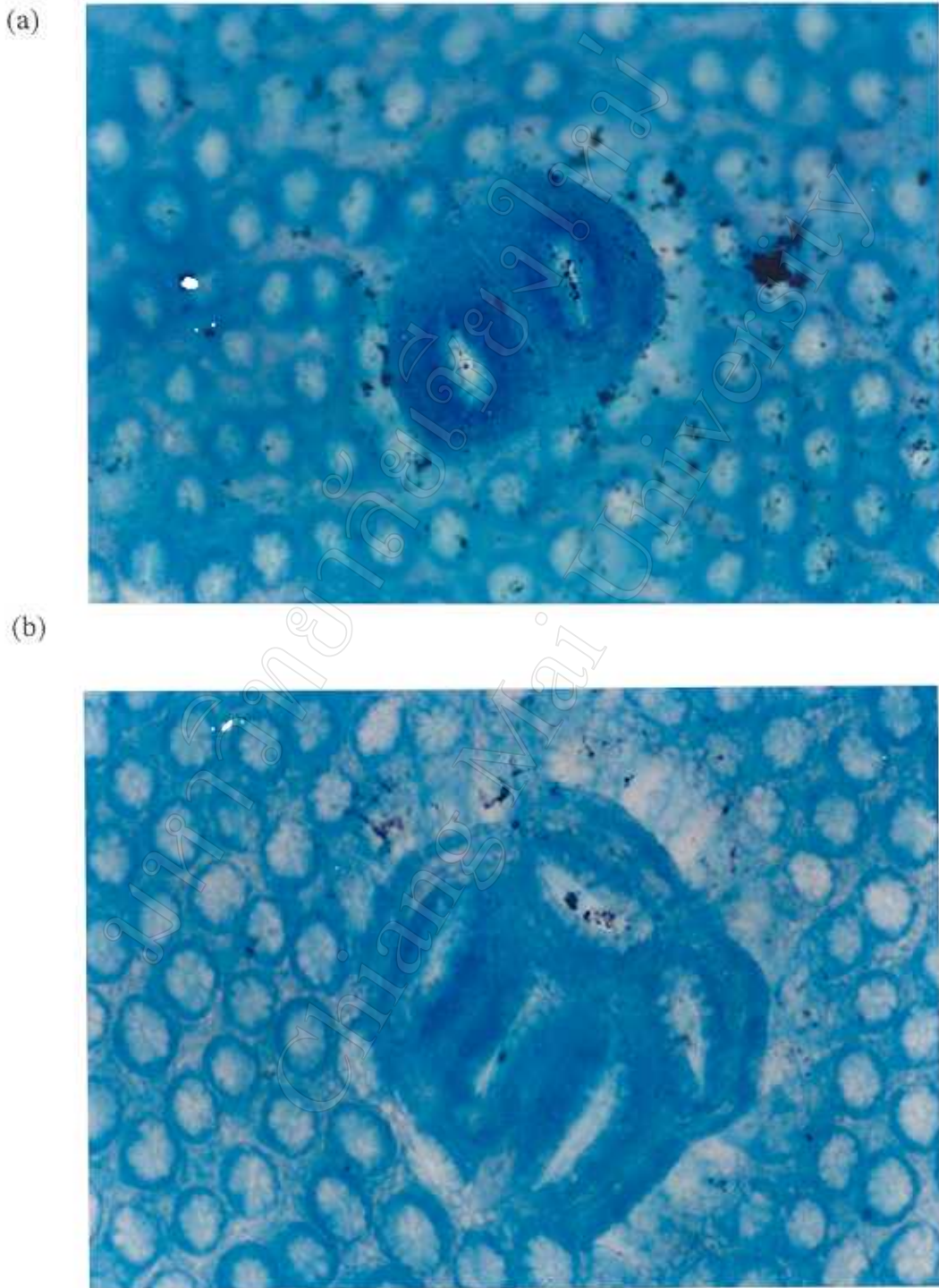
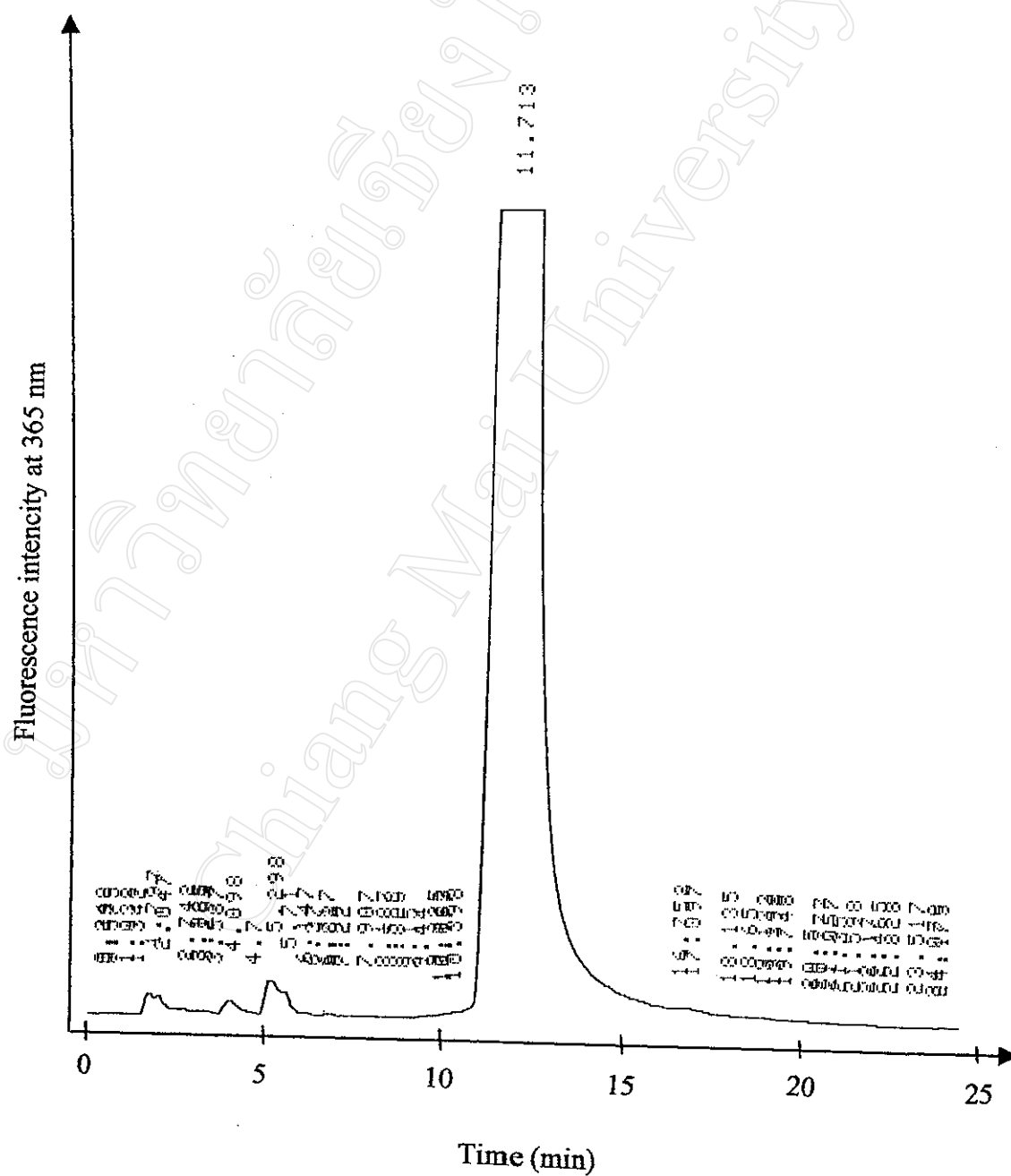


Figure 20. Morphology of Aberrant crypt foci (a) 2 crypts/focus and (b) more than 4 crypts/focus

Appendix 5. HPLC profile for DNA adduct determination

Figure 21. HPLC profile for standard O⁶-methylguanine

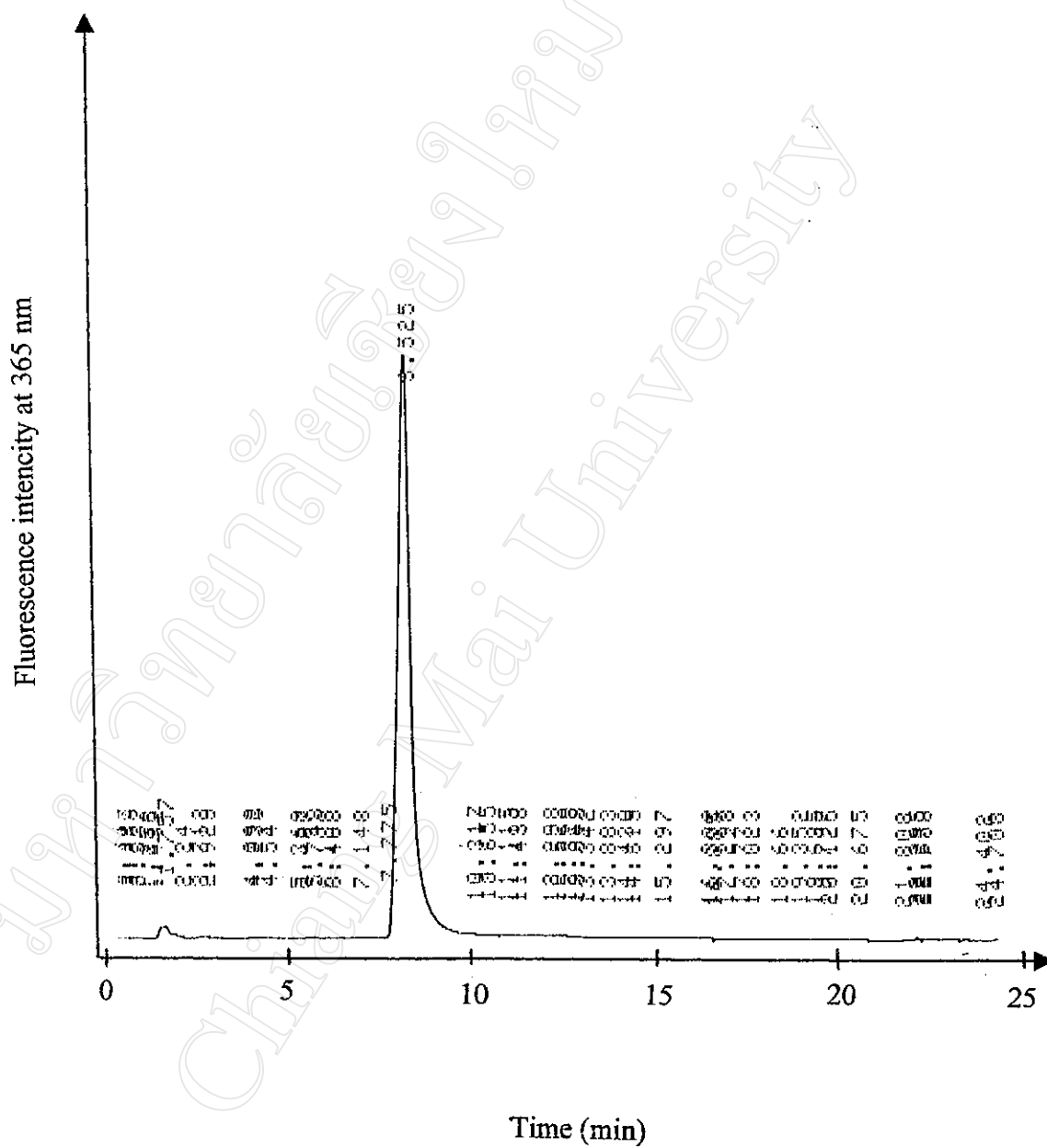


Figure 22. HPLC profile for standard N^7 -methylguanine

Figure 23. HPLC profile for DNA adduct sample

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Intiyot, Y., Kinouchi, T., Vinitketkumneun, U., Ohnishi, Y., (1997) Effect of *Murdannia loriformis* extract on cancer formation in rats colon. Presented at the 4th National Cancer Conference, 11-13 November, Bangkok, Thailand

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