

APPENDIX A

Calculations of F⁻ Content in Coal Samples

Ion chromatographic Technique

According to the procedure in Section 2.4.6.2-1(a), the original samples #20 referring to 0.9996 g of sample was extracted with DMSO/KNO₃ and adjusted with deionized water to the mark of a 50 ml volumetric flask. Then 1.0 ml was pipetted into 100 ml volumetric flask and diluted with deionized water to sample #20 solution.

$$\text{Weight of samples \#20 in sample solution} = 0.9996 \times 10^{-2} \text{ g/50 ml}$$

From calibration curve:

$$\text{Concentration of F}^{-} \text{ in sample solution} = 0.596 \text{ ng/}\mu\text{l}$$

$$\text{So, 50 ml of sample solution contained F}^{-} = 0.596 \times 50 \text{ }\mu\text{g}$$

$$= 2.98 \times 10^{-5} \text{ g}$$

$$\text{sample \#20 } 0.9996 \times 10^{-2} \text{ g contained F}^{-} = 2.98 \times 10^{-5} \text{ g}$$

$$\text{sample \#20 } 100 \text{ g contained F}^{-} = \frac{2.98 \times 10^{-5} \times 100}{0.9996 \times 10^{-2}} \text{ g}$$

$$= 0.30 \text{ g}$$

$$\therefore \text{F}^{-} \text{ content in sample \#20} = 0.30 \text{ \%w/w}$$

Calculations of Fe³⁺ Content in geological reference materials

Ion chromatographic Technique

Soil 5

According to the procedure in Section 2.5.7.2-Method II, the original samples #22 referring to 0.0994 g of sample was dissolved with 50 % HNO₃ and adjusted with deionized water to the mark of a 25 ml volumetric flask. Then the sample solution was diluted 20-fold with deionized water.

Weight of samples #22 in sample solution was

$$\begin{aligned} \text{diluted 20-fold} &= 0.0994/20 && \text{g} \\ &= 4.97 \times 10^{-3} \text{ g}/25 && \text{ml} \end{aligned}$$

From calibration curve:

$$\text{Concentration of Fe}^{3+} \text{ in sample solution} = 2.46 \quad \text{ng}/\mu\text{l}$$

$$\text{So, 50 ml of sample solution contained Fe}^{3+} = 2.46 \times 25 \quad \mu\text{g}$$

$$= 6.15 \times 10^{-5} \quad \text{g}$$

$$\text{sample \#22 } 4.97 \times 10^{-3} \text{ g contained Fe}^{3+} = 6.15 \times 10^{-5} \quad \text{g}$$

$$\text{sample \#22 } 100 \text{ g contained Fe}^{3+} = 6.15 \times 10^{-5} \times 100 \quad \text{g}$$

$$4.97 \times 10^{-3}$$

$$= 1.24 \quad \text{g}$$

$$\therefore \text{Fe}^{3+} \text{ content in sample \#22} = 1.24 \quad \% \text{w/w}$$

APPENDIX B

Radii in Angstrom units (A) of selected atoms and ions.

F	0.64	Cl	0.99	Br	1.14
F ⁻	1.36	Cl ⁻	1.81	Br ⁻	1.95

Mn	1.29	Fe	1.26	Cu	1.28	Pb	1.75
Mn ²⁺	0.80	Fe ²⁺	0.76	Cu ⁺	0.96	Pb ⁴⁺	0.84
Mn ³⁺	0.66	Fe ³⁺	0.64	Cu ²⁺	0.72	Pb ²⁺	1.32
Co	1.25	Ni	1.24	Zn	1.33		
Co ²⁺	0.74	Ni ²⁺	0.69	Zn ²⁺	0.74		

Source : Wall Chart entitled "Relative Sizes of Atoms and Ions in the Periodic Table", Sargent-Welch Scientific Co., 1970.

NO ₂ ⁻	1.55	SO ₄ ²⁻	2.30
NO ₃ ⁻	1.89	PO ₄ ³⁻	2.38

Source : Shriver, D.F., Atkins, P.W. and Longford, C.H., Inorganic Chemistry, Oxford University Press, Oxford, 1994, p.173.

APPENDIX C

Properties of Some Metal-PAR Species.

Table C-1 Properties of Some Metal-PAR Species^a

	Approx. pH	M:PAR	λ_{\max} (nm)	$\epsilon \times 10^{-3}$
Au(III)		1:1	540	8.3 (tert-butanol + xylene)
Bi	1	1:1	515	10.7
Cd	10	1:2	495	57.8
Co(III?)		1:2	510	5.5
Cu(II) ^b	2 ; 7 ; 10	1:1 (pH 2.3-5)	522	12.1
		1:2 (pH>5)	505-510	58.9
Ga	4 ; 7	1:1 (pH 1.5-3)	490-495	21.2
		1:2 (pH 3-5)	500-505	9.9
Hf		1:4 (pH2.5)	510	37.5 (isoamyl alcohol)
In	7	1:1	500-510	32.8
Mn	10	1:2	496	86.5
Nb	5-6	1:1 (0.1-0.2 N H ₂ SO ₄)	530	18
		1:1 (pH~6)	555	38.7 , 31.2
Pb	10	1:1	512	10.8
		1:2	522	50.2
Pd	6 ; H ₂ SO ₄ soln.	1:1 (H ₂ SO ₄ solution)	440	18.4
Pt(II)	3	1:1	450, 660	22.9
Rare earths	6-7	1:2	515	~16 to 50
Sc	3-5	1:1(pH 2)	505 (515)	14.7 (22.1)
Ta	5	1:1	515	20.4
Th	7	1:4	500	38.9

Table C-1 Properties of Some Metal-PAR Species^a (Cont.)

	Approx. pH	M:PAR	λ_{\max} (nm)	$\epsilon \times 10^{-3}$
Tl(III) ^c		1:1	520	18-19.4
U(VI)	7-8	1:1	510 (530)	38.5
V(V)	6.5	1:1	~550	
Zn	8		495	81
Zr		1:1 (pH 4)	535	21

Source : ^a Condensed from Busev and Ivanov, *loc. city.*, with later data from other authors. Solvent: H₂O.

^b O.A. Tataev, S.A. Akhmedov, and Kh.A. Akhmedova, *Zh. Anal. Khim.*, 24, 834(1969), report formation of a 1:1 complex at pH 1.5-2.5, $\epsilon_{540} = 27,000$.

^c As shown by E.A. Biryuk and R.V. Ravitskaya, *Zh. Anal. Khim.*, 26, 1767 (1971), Tl(III) forms Tl(OH)₂L (HL = PAR and PAN).

APPENDIX D

Table D-1 Values of t for ν Degree of Freedom for Various Confidence Levels

ν	Confidence Level, %	90	95	99	99.5
1		6.314	12.706	63.657	127.32
2		2.920	4.303	9.925	14.089
3		2.353	3.182	5.841	7.453
4		2.132	2.776	4.604	5.598
5		2.015	2.571	4.032	4.773
6		1.943	2.447	3.707	4.317
7		1.895	2.365	3.500	4.029
8		1.860	2.306	3.355	3.832
9		1.833	2.262	3.250	3.690
10		1.812	2.228	3.169	3.581
15		1.753	2.131	2.947	3.252
20		1.725	2.086	2.845	3.153
25		1.708	2.060	2.787	3.078
α		1.645	1.960	2.576	2.807

Source : Christian, G.D., Analytical Chemistry, 3rd ed., John Wiley & Sons, New York, 1980, p.71.

APPENDIX E**Detail about geological reference materials****Detail about geological reference materials in this work****Symbol**

- = precision of compiled data established to be better than 10% relative (two sigma) normally based on five or more results from two or more independent techniques.
- ? = additional uncertainty, caused, for example, by a large disparity in reported results, and/or a small number of reported results (usually < 3) and/or data derived from only one “non-definitive” analytical technique.
- calc = Value calculated from quoted data, normally assuming oxide stoichiometry
- cv = Datum quoted as a certified value by the distributing organization.

Soils

wt. %	GSS	GSS	GSS	GSS	IAEA	IAEA
	tropical soil CS-1	tropical soil K-3	tropical soil SAu-1	tropical soil SUR-1	soil SOIL-5	soil SOIL-7
SiO ₂	45.00	47.75	42.40	45.30	71 ?	38.5 ?
TiO ₂	1.06	1.83	1.80	0.93	0.78 ?	0.5 ?
Al ₂ O ₃	17.00	23.00	20.50	14.90	15.48	8.9 ?
Fe ₂ O ₃	-	-	-	-	-	-
FeO	-	-	-	-	-	-
MnO	0.05	0.04	0.10	0.04	0.110	0.0815 cv
MgO	0.19	0.05	0.11	0.10	2.5 ?	1.37 ?
CaO	0.05	0.02	0.04	0.03	3.1 ?	22.8 ?
Na ₂ O	0.10	0.10	0.04	0.01	2.59	0.32 ?
K ₂ O	0.50	0.03	0.05	0.11	2.24	1.46 ?
P ₂ O ₅	0.07	0.05	0.08	0.06	0.25 ?	0.105 ?
H ₂ O ⁺	-	-	-	-	-	-
H ₂ O ⁻	-	-	-	-	-	0 dried
CO ₂	-	-	-	-	-	-
C	-	-	-	-	-	-
SUM Tr.	0.19	0.20	0.27	0.23	0.33	0.16
Total	64.21	73.07	65.39	61.71	98.38	74.69
Fe ₂ O ₃ T	24.70	14.73	21.90	28.40	6.36	3.67 ?
L.O.I.	11.25	11.60	12.20	10.10	0 calc	0 calc
XTotal	99.97	99.20	99.22	99.98	104.41	78.21
REF	BUR85	BUR85	BUR85	BUR85	DYB79	IAEA84
ppm	CS-1	K-3	SAu-1	SUR-1	SOIL-5	SOIL-7
Ag	-	-	-	-	1.9 ?	-
As	8	2.3	4	11	94	13.4 cv
Au	0.33 ?	-	0.2 ?	0.3 ?	-	-
B	-	-	-	-	63 ?	-
Ba	220	27	50	45	562	159 ?
Be	-	0.2 ?	0.6 ?	0.3 ?	1.8 ?	-
Bi	0.6 ?	34 ?	8 ?	11 ?	12 ?	-
Br	5.9 ?	3.8 ?	6.6 ?	4.3 ?	5.4	7 ?
Cd	15 ?	-	16 ?	15 ?	1.5 ?	1.3 ?
Ce	27.7 pot91	56.4 pot91	18.8 pot91	13.5 pot91	59.7	61 cv
Cl	50 ?	12 ?	250 ?	-	-	-
Co	-	-	35	24	14.8	8.9 cv
Cr	500	200	640	500	28.9	60 cv
Cs	1.27 pot91	0.32 ?	0.6 pot91	0.9 pot91	56.7	5.4 cv
Cu	110	45	150	74	77.1	11 cv
Dy	-	-	-	-	4.0	3.9 cv
Er	-	-	-	-	-	-
Eu	0.63 pot91	0.56 pot91	0.50 pot91	0.37 pot91	1.18	1.0 cv
F	-	180 ?	-	50 ?	628 ?	480 ?
Ga	-	19 ?	-	-	18.4	10 ?
Gd	-	-	-	-	35 ?	-
Ge	-	-	-	-	2.3 ?	-

Soils

ppm	CS-1	K-3	SAu-1	SUR-1	SOIL-5	SOIL-7
Hf	4.55 pot91	15.5 pot91	3.8 pot91	3.58 pot91	6.30	5.1 cv
Hg	-	-	-	-	0.79 ?	0.04 ?
Ho	-	-	-	-	0.82 ?	1.1 cv
I	-	-	-	-	35 ?	-
In	-	2 ?	-	-	-	-
Ir	-	-	-	0.007 ?	-	-
La	12	16	6 ?	6 ?	28.1	28 cv
Li	18 ?	12 ?	2 ?	13 ?	52.1 ?	31 ?
Lu	0.23 pot91	0.26 pot91	5.6 pot91	6.0 pot91	0.336	0.3 ?
Mo	24	5	18 ?	-	1.7 ?	2.5 ?
N	-	-	-	-	-	-
Nb	-	-	-	-	9 ?	12 ?
Nd	11.4 pot91	13.4 pot91	6.5 pot91	-	29.9	30 cv
Ni	62	43	80	60	13 ?	26 ?
Os	-	-	-	-	-	-
Pb	17	17	12 ?	10 ?	129	60 cv
Pd	-	-	-	-	-	-
Pr	-	-	-	-	5.0 ?	-
Pt	-	-	-	0.15 ?	-	-
Ra Bq/g	-	-	-	-	-	-
Rb	7.3 ?	5.7 ?	9 ?	8 ?	138	51 cv
Rc	-	-	-	-	-	-
Rh	-	-	-	0.02 ?	-	-
Ru	-	-	-	-	-	-
S	-	500 ?	600 ?	600 ?	-	-
Sb	1 ?	-	0.9 ?	1 ?	14.3	1.7 cv
Sc	40.7 pot91	30.1 pot91	62.0 pot91	39.9 pot91	14.8	8.3 cv
Se	2.2 ?	1.5 ?	-	1.6 ?	1.3 ?	0.4 ?
Sm	2.18 pot91	2.58 pot91	1.58 pot91	1.31 pot91	5.42	5.1 cv
Sn	-	1.3 ?	-	5 ?	4.2 ?	-
Sr	23	10	9	7	330 ?	108 cv
Ta	0.64 pot91	1.14 pot91	0.53 pot91	0.49 pot91	0.764	0.8 cv
Tb	0.38 pot91	0.34 pot91	0.43 pot91	0.31 pot91	0.665	0.5 cv
Te	-	-	-	-	-	-
Th	8.3 pot91	9.9 pot91	3.3 pot91	6.9 pot91	11.3	8.2 cv
Tl	-	-	-	-	-	-
Tm	-	-	-	-	0.42 ?	-
U	0.9 pot91	1.5 pot91	1.26 pot91	1.9 pot91	3.15	2.6 cv
V	450	220	460	600	151 ?	66 cv
W	10	-	-	2 ?	5.1 ?	-
Y	9 ?	5 ?	8 ?	6 ?	21 ?	21 cv
Yb	1.35 pot91	1.62 pot91	1.32 pot91	1.04 pot91	2.24	2.4 cv
Zn	65	50	105	60	368	104 cv
Zr	150	500	115	110	221 ?	185 cv
REF	BUR85	BUR85	BUR85	BUR85	DYB79	IAEA84

Sediments

wt. %	BAS	BCR	BCR	BCR	BSRRM	BSRRM
	ball clay BCS 348	estuarine sediment CRM 277	lake sediment CRM 280	river sediment CRM 320	grey shale AWI-1	micaceous siltite PRI-1
SiO ₂	51.1 cv	-	-	-	60.35	68.63
TiO ₂	1.08 cv	-	-	-	0.91	0.72
Al ₂ O ₃	31.6 cv	-	-	-	16.55	10.89
Fe ₂ O ₃	1.04 cv	-	-	-	1.30	1.01
FeO	-	-	-	-	5.26	2.04
MnO	-	0.19 cv	0.17 cv	0.10 cv	0.14	0.04
MgO	0.30 cv	-	-	-	2.00	3.12
CaO	0.17 cv	-	-	-	0.67	2.44
Na ₂ O	0.34 cv	-	-	-	0.76	1.73
K ₂ O	2.23 cv	-	-	-	3.07	3.77
P ₂ O ₅	0.071 cv	-	-	-	0.15	0.19
H ₂ O ⁺	-	-	-	-	-	-
H ₂ O ⁻	0 dried	-	-	-	0 dried	0 dried
CO ₂	-	-	-	-	1.46 ?	2.62 ?
C	1.64 T	-	-	-	-	-
SUM Tr.	0.07	0.17	0.16	0.14	0.33	0.22
Total	89.66	-	-	-	92.95	97.42
Fe ₂ O ₃ T	1.04 cv	-	-	-	7.14	3.28
L.O.I.	11.8 cv	-	-	-	7.70	4.99
XTotal	99.75	-	-	-	99.44	99.80
REF	BAS89	BCR	BCR	BCR	ROE88	ROE88

ppm	BCS 348	CRM 277	CRM 280	CRM 320	AWI-1	PRI-1
Ag	-	-	-	-	-	-
As	-	47.3 cv	51.0 cv	76.7 cv	15.9	13 ?
Au	-	-	-	-	-	-
B	-	-	-	-	161	54 ?
Ba	360 ?	329 ?	617 ?	531 cv	384	510
Be	-	-	-	-	-	-
Bi	-	-	-	-	-	-
Br	-	87 ?	11 ?	-	-	-
Cd	-	11.9 cv	1.6 cv	0.53 cv	-	-
Ce	-	-	70 ?	95 ?	82.4	79
Cl	-	-	-	-	-	-
Co	-	17 ?	20 ?	19 ?	21.2	6.5
Cr	110	192 cv	114 cv	138 cv	117	76
Cs	-	-	-	-	6.8 ?	2.2 ?
Cu	-	101.7 cv	70.5 cv	44.1 cv	36	3 ?
Dy	-	-	-	-	4.2	3.5
Er	-	-	-	-	-	-
Eu	-	-	-	-	153	1.28
F	-	-	-	-	577 ?	394 ?
Ga	-	-	-	-	24 ?	18 ?
Ge	-	-	-	-	5.6	4.9
Gr	-	-	-	-	-	-

Sediments

ppm	BCS 348	CRM 277	CRM 280	CRM 320	AWI-1	PRI-1
Hf	-	-	-	-	6.12	10.7
Hg	-	1.77 cv	0.670 cv	1.03 cv	-	-
Ho	-	-	-	-	0.9 ?	1.0 ?
I	-	-	-	-	-	-
In	-	-	-	-	-	-
Ir	-	-	-	-	-	-
La	-	45 ?	39 ?	46 ?	38.2	37.3
Li	-	-	-	-	-	-
Lu	-	-	-	-	0.46	0.39
Mo	-	-	-	-	-	-
N	-	-	-	-	-	-
Nb	-	-	-	-	-	-
Nd	-	-	-	-	37.5	37.6
Ni	-	43.4 cv	73.6 cv	75.2 cv	66	22
Os	-	-	-	-	-	-
Pb	-	146 cv	80.2 cv	42.3 cv	22	12
Pd	-	-	-	-	-	-
Pr	-	-	-	-	-	-
Pt	-	-	-	-	-	-
Ra Bq/g	-	-	-	-	-	-
Rb	-	-	-	-	130	88
Re	-	-	-	-	-	-
Rh	-	-	-	-	-	-
Ru	-	-	-	-	-	-
S	-	-	-	-	869	170 ?
Sb	-	4.0 ?	1.4 ?	0.6 ?	0.74 ?	0.24 ?
Sc	-	9.0 cv	12.8 cv	15.25 cv	16.3	9.49
Se	-	2.04 cv	0.68 cv	0.214 cv	-	-
Sm	-	-	-	-	7.1	6.6
Sn	-	-	-	-	7 ?	2 ?
Sr	-	-	-	-	107	87
Ta	-	-	-	-	1.37	1.06
Tb	-	-	-	-	0.91	0.77
Tc	-	-	-	-	-	-
Th	-	9.0 ?	15 ?	18 ?	12.2	12.1
Ti	-	-	0.7 ?	0.5 ?	-	-
Tm	-	-	-	-	0.18 ?	0.29 ?
U	-	-	-	-	3.1	2.49
V	-	102 ?	101 ?	105 ?	141	72
W	-	-	-	-	-	-
Y	-	-	-	-	32 ?	21 ?
Yb	-	-	-	-	3.00	2.40
Zn	-	547 cv	291 cv	142 cv	100	41
Zr	220 ?	-	-	-	223	382
REF	BAS89	BCR	BCR	BCR	ROES8	ROE88

Reference to source data.

fug 85 = R. Fuge (1985)

ter 84 = S. Terishima (1986)

vic 86 = A.H. Victor (1986)

Cro 82 = I. Croudace, J.-L. Joron, H. Jaffrezic, G. Meyer and Treuil (1982)

ran 89 = K.R. andle and I.W. Croudace (1989)

roe 85 = I. Roelandts, G. Robaye, G. Weber and J.M. Delbrouck (1985)

els 87 = H.N. Elsheimer (1987)

vic 80 = A.H. Victor and F.E.W. Strelow (1980)

Source : P.J. Potts, A.G. Tindle and P.C. Webb, *Geochemical Reference Material Composition*, Whittles, London, 1992, pp.(54-55) and (84-85).

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