

REFERENCES

1. Holum, J. R., "Introduction to Organic and Biological Chemistry", John Wiley & Sons, New York, 1969, pp. 219-220.
2. <http://www.scifun.chem.wisc.edu/chemweek/ETHANOL/ethanol.html>, [August 20, 2006].
3. <http://www.en.wikipedia.org/wiki/Ethanol>, [August 24, 2006].
4. Kocache, R., Gas Sensors, *Sensor Rev.*, **14**, 1994, 8-12.
5. Liu, C. C., Hesketh, P. J., and Hunter, G. W., Chemical Microsensors, *Electrochem. Soc. Interface.*, 2004, 22-27.
6. Stetter, J. R., Hesketh, P. J., and Hunter, G. W., Sensors: Engineering Structures and Materials from Micro to Nano, *Electrochem. Soc. Interface.*, 2006, 66-69.
7. <http://www.delphian.com/electrochemical%20sensors.htm>, [August 24, 2006].
8. http://www.nemoto.co.jp/column/09_ecco_e.html, [August 26, 2006].
9. Capone, S., Forleo, A., Francioso, L., Rella, R., Siciliano, P., Spadavecchia, J., Presicce, D. S., and Taurino, A. M., Solid State Gas Sensors: State of the Art and Future Activities, *J. Electron. Mater.*, **5**, 2003, 1335 – 1348.
10. Kong, X., and Wang, Z. L., Integration of metal oxide nanobelts with Microsystems for nerve agent detection, *Appl. Phys. Lett.* **86** (2005) 063101-3.
11. Jie, Z., Hua, H. L., Shao, G., Hui, Z., and Gui, Z. J., Alcohols and acetone sensing properties of SnO₂ thin films deposited by dip-coating, *Sens. Actuators B: Chem.* **115**, 2006, 460-466.

12. Tianshu, Z., Hing, P., Li, Y., and Jiancheng Z., Selective detection of ethanol vapor and hydrogen using Cd-doped SnO₂-based sensors, *Sens. Actuators B: Chem.* **60**, 1999, 208-215.
13. Simakov, V., Yakusheva, O., Grebennikov, A., and Kisin, V., I-V characteristics of gas-sensitive structures based on tin oxide thin films, *Sens. Actuators B: Chem.* **116**, 2006, 221-225.
14. Montmeat, P., Lalauze1, R., Viricelle, J. P., Tournier, G., and Pijolat, C., Model of the thickness effect of SnO₂ thick film on the detection properties, *Sens. Actuators B: Chem.* **103**, 2004, 84-90.
15. Mandayo, G. G., Castano, E., Gracia, F. J., and Cirera, A., Strategies to enhance the carbon monoxide sensitivity of tin oxide thin films, *Sens. Actuators B: Chem.* **95**, 2003, 90-96.
16. Lettieri, S., Bismuto, A., Maddalena, P., Baratto, C., and Comini, E., Gas sensitive light emission properties of oxide and zinc oxide nanobelts, *J. Non-Cryst. Solids*, **352**, 2006, 1457-1460.
17. Zhu, B. L., Xie, C. S., Wang, W. Y., Huang, K. J., and Hu, J. H., Improvement in gas sensitive of ZnO thick film to volatile organic compounds (VOCs) by adding TiO₂, *Mater. Lett.*, **58**, 2004, 624-629.
18. Anno, Y., Maekawa, T., Tamaki, J. Y., Hayashi, A. K., Miura, N., and Yamazoe, N., Zinc-oxide-based semiconductor sensors for detecting acetone and capronaldehyde in the vapour of consomme soup, *Sens. Actuators B: Chem.* **25**, 1995, 623-627.

19. Cheng, X. L., Zhao, H., Huo, L. H., Gao, S., and Zhao, J. G., ZnO nanoparticulate thin film: preparation, characterization and gas-sensing property, *Sens. Actuator B: Chem.* **102**, 2004, 248-252.
20. Zhua, L., Xieb, C. S., Wangb, A. H., Zengb, D. W., Songb, W. L., and Zhao, X. Z., The gas-sensing properties of thick film based on tetrapod-shaped ZnO nanopowders, *Mater. Lett.*, **59**, 2005, 1004– 1007.
21. Kotsikau , D., Ivanovskaya , M., Orlik, D., and Falasconi, M., Gas-sensitive properties of thin and thick film sensors based on Fe₂O₃–SnO₂ nanocomposites, *Sens. Actuators B: Chem.* **101**, 2004, 199–206.
22. Tan, O. K., Cao, W., Zhu, W., Chai, J. W., and Pan, J. S., Ethanol sensors based on nano-sized α -Fe₂O₃ with SnO₂, ZrO₂, TiO₂ solid solutions, *Sens. Actuators B: Chem.* **93**, 2003, 396-401.
23. Reddy, B.V., Rasouli, F., Hajaligol, M. R., and Khanna, S. N., The NO₂ response of solid electrolyte sensors made using nano-sized LaFeO₃ electrodes, *Sens. Actuators B: Chem.* **76**, 2001, 483-488.
24. Zhao, S., Johnny, K. O., Sin, B. X., Zhao, M., Peng, Z., and Cai, Z., A high performance ethanol sensor based on field-effect transistor using a LaFeO₃ nano-crystalline thin-film as a gate electrode, *Sens. Actuators B: Chem.* **64**, 2000, 83-87.
25. Toan, N. N., Saukko ,S., and Lantto V., Gas sensing with semiconducting perovskite oxide LaFeO₃, *Physica*, **B 327**, 2003, 279-282.

26. Korotcenkov, G., Brinzari, V., Cerneavschi, A., Ivanov, M., Cornet, A., Morante, J., Cabot, A., and Arbiol, J., In₂O₃ films deposited by spray pyrolysis: gas response to reducing (CO, H₂) gases, *Sens. Actuators B: Chem.* **98**, 2004, 122-129.
27. Korotcenkov, G., Cerneavschi, A., Brinzari, V., Vasiliev, A., Ivanov, M., Cornet, A., Morante, J., Cabot, A., and Arbiol, J., In₂O₃ films deposited by spray pyrolysis as a material for ozone gas sensors, *Sens. Actuators B: Chem.* **99**, 2004, 297-303.
28. Korotcenkov, G., Brinzari, V., Cerneavschi, A., Ivanov, M., Golovanov, V., Cornet, A., Morante, J., Cabot, A., and Arbiol, J., The influence of film structure on In₂O₃ gas response, *Thin Solid Films*, **460**, 2004, 315-323.
29. Ionescu, R., Hoel, A., Granqvist, C. G., Llobet, E., and Heszler, P., Low-level detection of ethanol and H₂S with temperature-modulated WO₃ nanoparticle gas sensors, *Sens. Actuators B: Chem.* **104**, 2005, 132-139.
30. Guérin, J., Aguir, K., Bendahan, M., and Mauriat, C. L., Thermal modelling of a WO₃ ozone sensor response, *Sens. Actuators B: Chem.* **104**, 2005, 289-293.
31. Blo, M., Carotta, M. C., Galliera, S., Gherardi, S., Giberti, A., Guidi, V., Malagù, C., Martinelli, G., Sacerdoti, M., Vendemiati B., and Zanni, A., Synthesis of pure and loaded powders of WO₃ for NO₂ detection through thick film technology, *Sens. Actuators B: Chem.* **103**, 2004, 213-218.
32. Gong, H., Wang, Y. J., Teo, S. C., Huang, L., Interaction between thin-film tin oxide gas sensor and five organic vapors, *Sens. Actuators B: Chem.* **54**, 1999, 232-235.

33. Varghese, K. O., Malhotra, L. K., and Sharma, G. L., High ethanol sensitive in sol-gel derived SnO₂ thin films, *Sens. Actuators B: Chem.* **55**, 1999, 161-165.
34. Lee, D. S., Kim, Y. T., Huh, J. S., Lee, D. D., Fabrication and characteristics of SnO₂ gas sensor array for volatile organic compounds recognition, *Thin Solid Films*, **416**, 2002, 271-278.
35. Wanekaya, A. K., Uematsu, M., Breimer, M., and Sadik, O. A., Multicomponent analysis of alcohol vapors using integrated gas chromatography with sensor arrays, *Sens. Actuators B: Chem.* **110**, 2005, 41-48.
36. TGS 822 for the detection of Organic Solvent Vapors. Figaro Company Limited, USA.
37. Raymond, P. W., "Introduction to Analytical Gas Chromatography", 2nd Ed. Marcel Dekker, New York, 1998, pp. 9-11.
38. <http://www.lcresources.com/resources/getstart/3a01.htm>, [August 30, 2006].
39. Braithwaite, A., and Smith, F.J., "Chromatographic Methods", 5th ed., Blachie Academic & Professional, New York, 1996, pp. 48-51.
40. Francis, R., and Annick, R., "Chemical Analysis : Modern Instrumentation Methods and Techniques", 8th ed., John Wiley & Sons, New York, 2000, pp. 14-17, 23-32.
41. <http://www.chem.agilent.com/cag/peak/article10.html>, [August 30, 2006].
42. <http://openchemist.net/chemistry/show.php?id=analytical&story=instr001>, [August 30, 2006].

43. <http://www.chemistry.adelaide.edu.au/external/soc-rel/content/gc-col.htm>,
[August 30, 2006].
44. Tian, F., Simon, X. Y., and Kevin, D., Circuit and Noise Analysis of Odorant Gas Sensors in an E-Nose, *Sensors*, **5**, 2005, 85-96.
45. Zhao, Y., Feng, Z., and Liang, Y., SnO₂ gas sensor films deposited by pulsed ablation, *Sens. Actuators B: Chem.* **56**, 1999, 224-227.
46. Jiao, Z., Gang, Y., Chen, F., Li, M., and Liu, J., The preparation of ZnGa₂O₄ nano cyystals by spray coprecipitaton and its gas sensitive characteristics, *Sensors*, **2**, 2002, 71-78.
47. Ho, J. J., Fang, Y. K., Wu, K. H., Hsieh, W. T., Chen, C. H., Chen, G. S., Ju, M. S., Lin, J. J., and Hwang, S. B., High sensitivity ethanol gas sensor integrated with a solid-state heater and thermal isolation improvement structure for legal drink-drive limit detecting, *Sens. Actuators B: Chem.* **50**, 1998, 227-233.
48. Burgess, C., " Vaild Analytical Methods and Procedures ", Royal Society of Chemistry, Cambridge, 2000, pp. 9-12, 14.
49. Volker, T., Debbie, S., and David M., Limits of Detection in Spectroscopy, *Spectroscopy*, **18 (12)**, 2003, 122-124.
50. Miller, J. C., and Miller, J. N., " Statistics for Analytical Chemistry ", 3th ed. Simon& Schuster, New York, 1993, pp. 115-118.

APPENDIX A

1. Percent of relative standard deviation (% RSD)

Percent of relative standard deviation can be calculated from the following equation.⁴⁸

$$\% \text{ RSD} = (SD / \bar{X}) * 100$$

Where

% RSD = percent of relative standard deviation

SD = standard deviation

\bar{X} = mean measured value

SD can be calculated from the following equation :

$$SD = \sqrt{\frac{\sum_{i=1}^n (x_i - \bar{x})^2}{n-1}}$$

Where

x_i = individual measured

n = number of measurements

ลิขสิทธิ์มหาวิทยาลัยเชียงใหม่
Copyright© by Chiang Mai University
All rights reserved

APPENDIX B

A definition of limit of detection (LOD) is based on the concentration, which gives a signal equals to the blank signal plus three standard deviation of the blank.⁴⁹⁻⁵⁰ LOD is calculated from the calibration curve by means of the blank signal, which can be used as an estimation of the calculated intercept, plus three standard deviations of the blank. It can be used as an estimation of the calculated value from the regression line.

The limit of detection is calculated from the linear regression line of the calibration curve as :

$$y = bx + a \quad (1)$$

where

y = instrument signals

x = concentrations

a = intercept

b = slope of the straight line

$$Y_L = Y_B + kS_B \quad (2)$$

where

Y_L = lowest detectable instrument signals

Y_B = Blank signal

$Y_B \cong$ intercept, a