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ABBREVIATIONS AND SYMBOLS

A	Absorption
A ⁺	Electron acceptor
at.%	Atomic %
В	Peak width measured at half height measured in
	radius
BET	Brunauer-Emmett-Teller
С	Amount of carbon
C	A constant, related to the free energy of adsorption
c	Speed of light
CRT	Cathode-ray tube
CMU	Chiang Mai University
CA	Concentration of element A
	Concentration of element B
°Cyright ^O hy	Degrees Celsius
d _{hkl}	Interplanar distance between (hkl) planes
la righ	Lattice planar spacing S C V C O
cb	Conduction band
e	Electron
e _{cb}	Conduction band electron
eV	Electron volt

E	Binding energy
EDS, EDX	Energy dispersive X-ray spectroscopy
EM	Electron microscope
E_0	Energy of ground state
E_1	Energy of first excited state
Ea	Apparent activation energy
E _b	Binding energy
Eg	Optical band gap of the semiconductor
Ek	Kinetic energy
E_{vac}	Energy of vacuum level
g/L	grams/liter
h 500	Plank's constant $(6.63 \times 10^{-34} \text{ Js})$
hν	Photon energy
\mathbf{h}^+	Hole
h^+_{vb}	Valence band hole
I_0	Intensity of the incident beam
Ι	Intensity of the transmittance
I _A	Background subtracted peak intensities for A
I _B	Background subtracted peak intensities for B
IUPAC	International union of pure and applied chemistry
J	Intensity of the reflected radiation
JCPDS	Joint committee powder diffraction standards
K	Kelvin
Coxvright [©]	Absorption coefficient
k	Conductivity value
A keV	kilo electron volt
kV	kilo-volt
Ĩ	Wave vector
$ ilde{k}'_{ ext{cb}}$	Wave vector of the lowest energy state in the
	conduction band

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$ ilde{ m k}'_{ m vb}$	Wave vector of the highest energy state in the
	valence band
mg	Milligram
min	Minute
ml	Milliliter
mS	Millisiemen
n	Order of diffraction
nm	Nanometer (10 ⁻⁹ m)
Na	Avogadro's number (6.02×10^{23})
0	Oxygen
O_2^{\bullet}	Superoxide radical
OH•	Hydroxyl radical
Р	Pressure at the constant temperature
P_0	Saturation pressure at the measurement temperature
R_{lpha}	Absolute remittance
rpm	Revolution per minute
rk	Kelvin radius
rp	Actual pore radius
S	Twice the scattering coefficient of sample
SEM	Scanning electron microscopy
SSA	Specific surface area
S.A.	Surface area
S _{BET}	BET surface area
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TTIP	Titanium tetraisopropoxide
TEM	Transmission electron microscopy
t	Thickness (t) of adsorbed N ₂ layers
$t_{ m hkl}$	Particle size measured from X-rays diffracted from
	the (<i>hkl</i>) planes
UNSW	University of New South Wales

UV-Vis	Ultraviolet-visible
UV-Vis DRS	Ultraviolet-visible diffuse reflectance spectroscopy
V	Volume, reduced to standard conditions (STP) of
	gas adsorbed per unit mass of adsorbent at a given
	pressure
vb	Valence band
Vm	Volume of gas adsorbed at STP per unit mass of
	adsorbent, when the surface is covered by a
	unimolecule layer of adsorbate
$V_{ m mol}$	Molar volume of absorbate gas at STP (22.4 mol ⁻¹)
WDS	Wavelength dispersive X-ray spectroscopy
XPS	X-ray photoelectron spectroscopy
XRD	X-ray diffraction
Z	Atomic number
λ	Wavelength
μg	Microgram (10 ⁻⁶ g)
μg C	Microgram of carbon
μm	Micron (10 ⁻⁶ meter)
μS/cm	MicroSiemens /square centimeter
Φ	Work function
3	Absorptivity
Hansurg	Bragg angle for the reflection
v	Frequency
v _{as} nght by	Frequency asymmetric
v _s	Frequency symmetric

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