

ลิขสิทธิ์มหาวิทยาลัยเชียงใหม่

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APPENDIX A

a-C:Mo Thin Film

Feb 14

8.81 k 35.1 k 2.767 k

> 500 MΩ

> 500 Mis

> 500 MΩ

> 500 MΩ

> 200 MΩ

34.2 k 217.0

242.8 88

> 2415 34.3 k 2159

> 238.5 34.1 k 212.7

2342 33.5 k 2069

230.0

12 Ξ

30×104

Jan 24,2006 Jan 24,2006

Ex 015

34.1 k 238.0

> 31.90 k 206.04 > 500 MΩ

> > 13×10*

Jan 25,2006

48×10

8 255

Ex 016 Ex 017 Ex 018

13×104

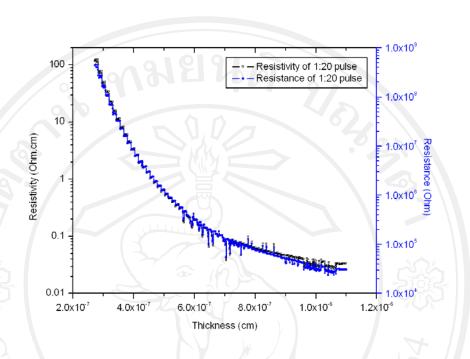
Jan 26,2006

212.1

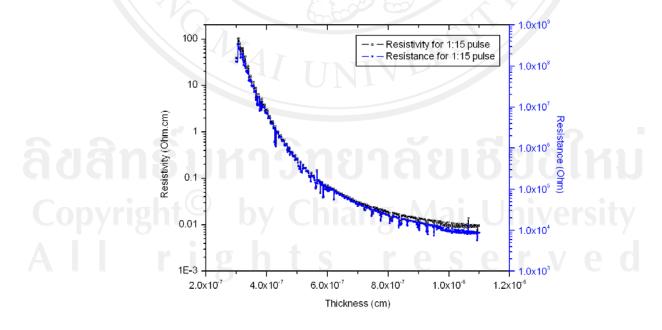
2/62/31 Appendix A-1: Additional data of a-C:Mo experiment.

Additional data for Molybdenum doped amorphous carbon film deposited on micro glass slide. Each sample deposited with 2,000 arc total pulse. Sheet resistance were measured by two ferminal configuration technique. Sheet resistance with 2,000 arc total pulse. Sheet resistance were measured by two ferminal configuration technique. This has been seed to the processed of the processed	ai		10				
Additional data for Molybde num doped amorphous carbon film deposited on micro glass slide. Each sample deposited with 2,000 Sheet resistance were measured by two terminal configuration technique. This lass slide. Each sample deposited with 2,000 Sheet resistance (3) Sample Mo.C pulse ratio Date processed Date processed Date processed Date processed Date of Date of Date Date of	arc total puls	(8)	Feb 8	8.80 k	35.2 k	2.769 k	637
Additional data for Molybdenum doped amorphous carbon film deposited on micro glass slide. Each sample depo Each training configuration technique. At the date and processed (Tox) Pressure (Tox) This land (Tox) At the date and processed (Tox) At the date and processed (Tox) At the date and processed (Tox) Sheet mail Sheet mail Ex.011 1:15 1an.20,2006 13 x 10¢ 11 844 k 866 k 872 k Ex.012 1:30 1an.20,2006 24 x 10¢ 13 32.34 k 34.1 k 34.6 k Ex.013 1:30 1an.20,2006 24 x 10¢ 13 2723 2.74 k 2.75 k Ex.013 1:5 1an.20,2006 13 x 10¢ 11 622 627 632	sited with 2,000	stame (Ω)	Jan 31	8.73 k	34.7 k	2.761 k	259
Additional data for Molybdenum doped amorphous carbon film deposited on micro glass slide. Eac Sheet resistance were measured by two ferminal configuration technique. This laws a law	h sample depo	Sheet resi	Jan 30	8.72k	34.6 k	2.738 k	632
Additional data for Molybdenum doped amorphous carbon film deposited on micro ging Seet resistance were measured by two ferminal configuration technique. This lass of two ferminal configuration technique. Sample Mo.C pube ratio Date processed Pressure (Tox) This lass of two ferminal configuration technique. Ex.011 1.15 Jan.20,2006 13 x 10^4 11 844 k Ex.012 1.20 Jan.20,2006 24 x 10^4 11 32.24 k Ex.013 1.10 Jan.20,2006 24 x 10^4 13 27.23 Ex.014 1.5 Jan.24,2006 13 x 10^4 11 622	ass slide. Eac	48	Jan 25	8.66k	34.1 k	2.74 k	623
Additional data for Molybdenum doped amorphous carbon film deposit Sheet resistance were measured by two terminal configuration technic fample Mo.C pulse ratio Date processed Pressure (Ton) Thichess (Ton) Ex.011 1.15 Jan.20,2006 13 x 104 11 Ex.012 1.10 Jan.20,2006 24 x 104 11 Ex.013 1.10 Jan.20,2006 13 x 104 11 11 11 12 x 104 11 11 11 11 11 11 11 11 11 11 11 11 11	ed on micro gl lue.		Atthe date processed	8.44 k	32.24 k	2723	622
Additional data for Molybdenum doped amorphous carb Sheet resistance were measured by two terminal config Sample Mo.C pulse ratio Date processed Pressure (Tox) Ex.011 1.15 Jan.20,2006 13 x 10⁴ Ex.012 1.20 Jan.20,2006 24 x 10⁴ Ex.013 1.10 Jan.23,2006 24 x 10⁴ Ex.014 1.5 Jan.24,2006 13 x 10⁴	on film deposit uration technic	F	(umx)	11	111	13	-11
Additional data for Molybdenum doped a Sheet resistance were measured by two Sample Mo.C pulse ratio Date processed Ex.011 1.15 Jan.20,2006 Ex.012 1.20 Jan.20,2006 Ex.013 1.10 Jan.29,2006 Ex.014 1.5 Jan.24,2006	morphous carb terminal config		Pressure (Torr)	13×104	2.4 x 10 ⁴	2.4 x 104	13×104
Additional data for Molyb Sheet resistance were m Sample Mo.C pube ratio Ex.011 1.15 Ex.012 1.20 Ex.013 1.10 Ex.014 1.5	denum doped a easured by two		Date processed	Jan 20,2006	Jan 20,2006	Jan 23,2006	Jan 24,2006
Addition: Sheet re Sample Ex 011 Ex 012 Ex 013	al data for Molyb. sistance were m		Mo:C pulse ratio	1:15	1:20	1:10	1.5
	Additions Sheet re:			Ex 011	Ex 012	Ex 013	Ex 014

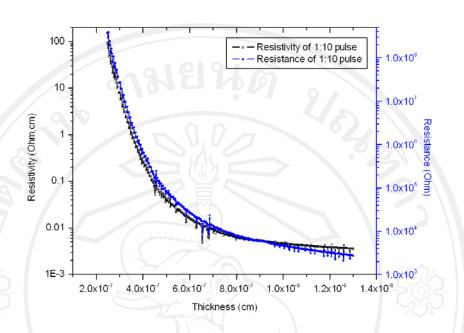
Appendix A-2: Film resistance during deposition



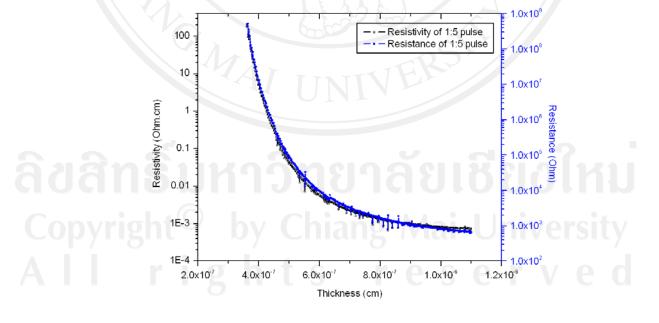
Resistance and resistivity during deposition of Sample Ex 012 (no bias).



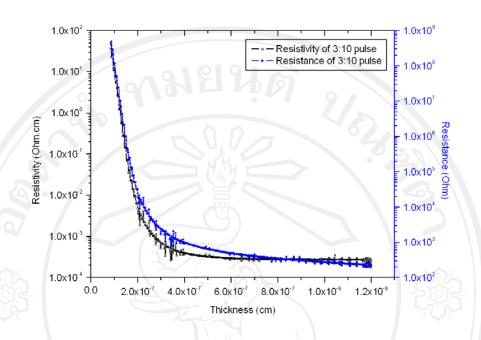
Resistance and resistivity during deposition of Sample Ex 011 (no bias).



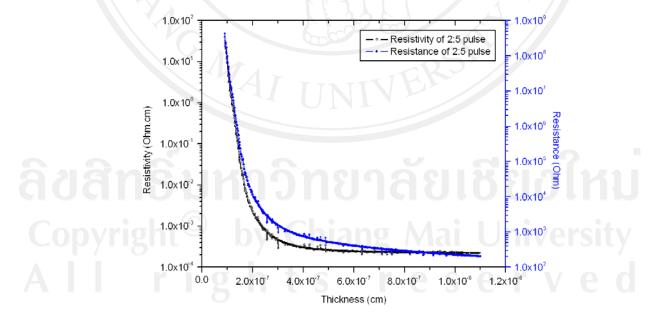
Resistance and resistivity during deposition of Sample Ex 013 (no bias).



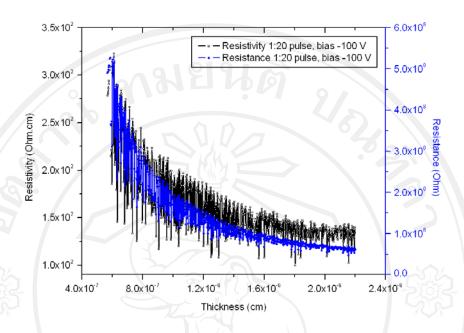
Resistance and resistivity during deposition of Sample Ex 014 (no bias).



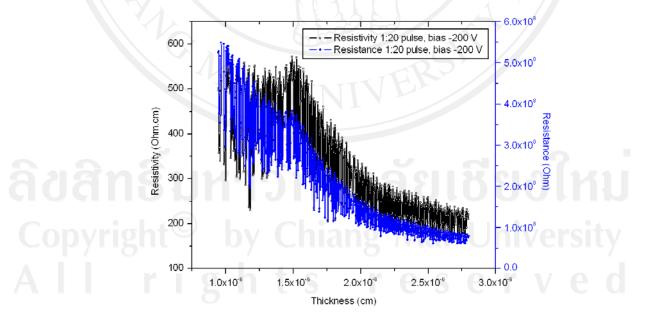
Resistance and resistivity during deposition of Sample Ex 015 (no bias).



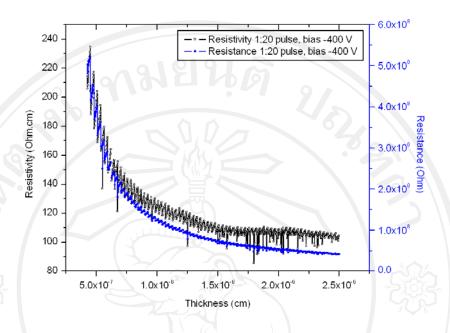
Resistance and resistivity during deposition of Sample Ex 017 (no bias).



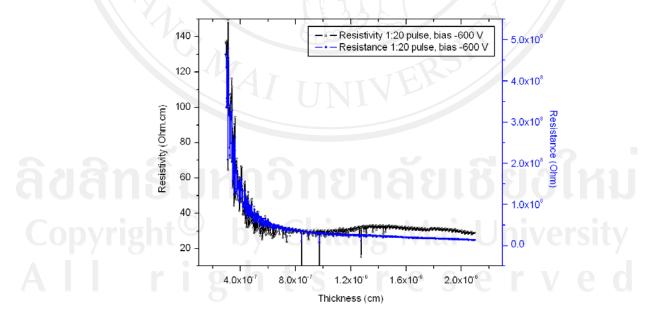
Resistance and resistivity during deposition of Sample Ex 037.



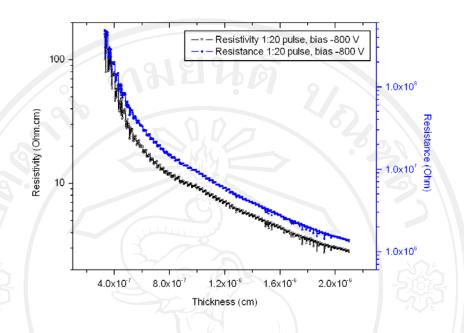
Resistance and resistivity during deposition of Sample Ex 038.



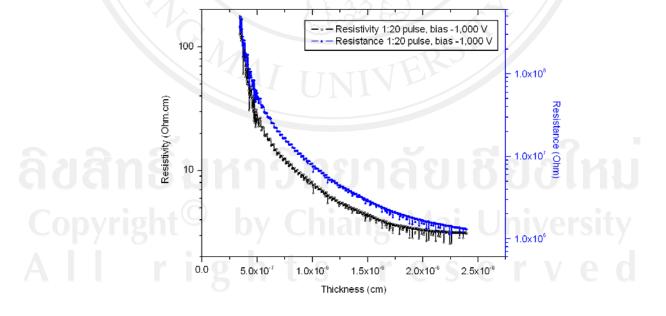
Resistance and resistivity during deposition of Sample Ex 039.



Resistance and resistivity during deposition of Sample Ex 040.



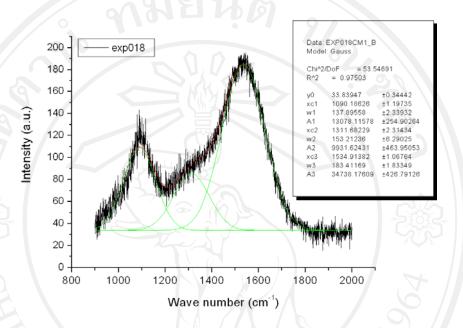
Resistance and resistivity during deposition of Sample Ex 041.



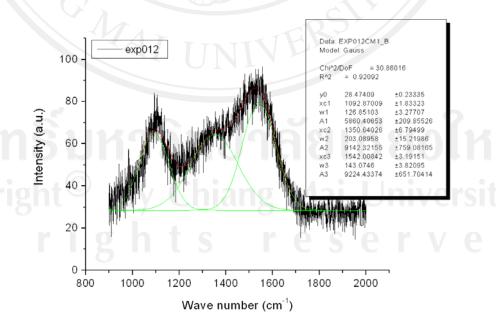
Resistance and resistivity during deposition of Sample Ex 042.

Appendix A-3: Raman spectra Gaussian fit.

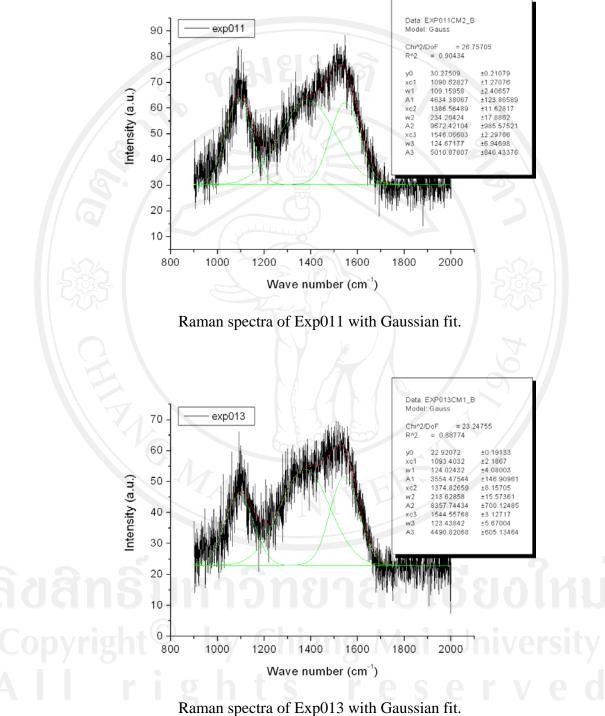
There are three peaks in the Raman spectra ; SiO_2 from substrate, D-peak and G-peak from the diamond like carbon film.

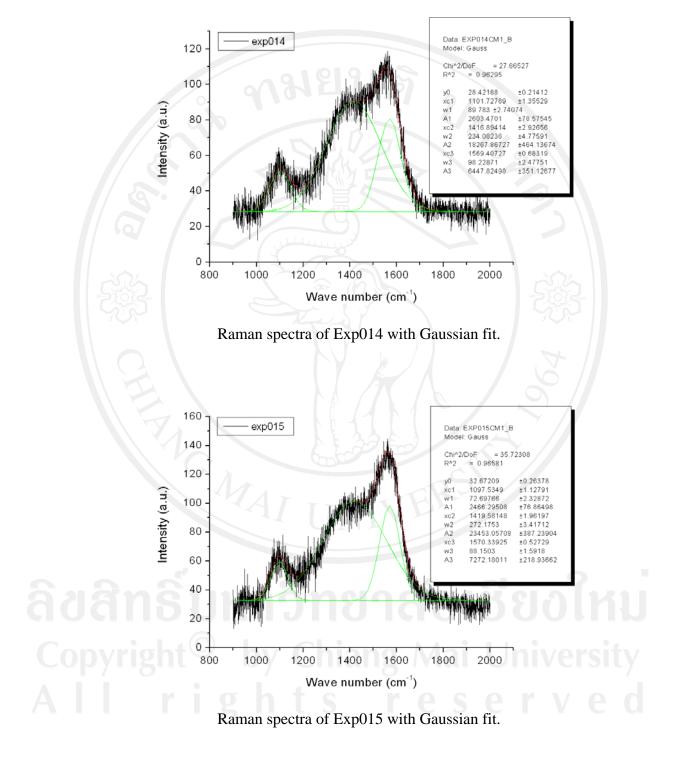


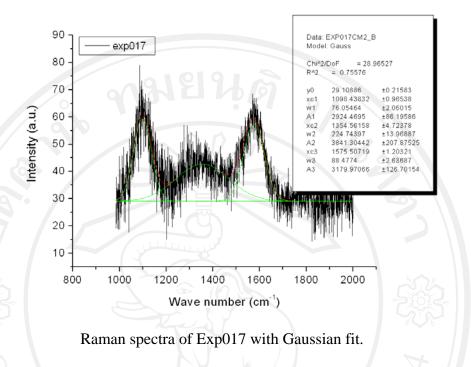
Raman spectra of Exp018 with Gaussian fit.

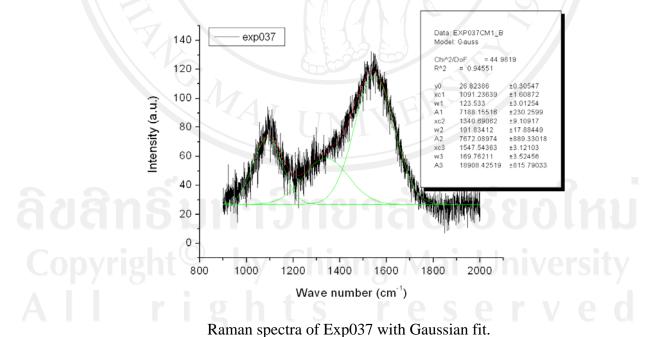


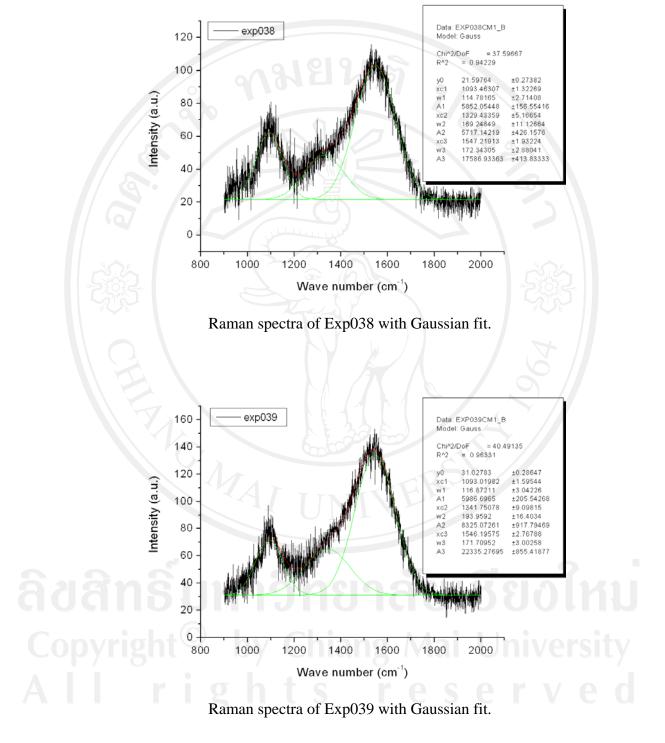
Raman spectra of Exp012 with Gaussian fit.

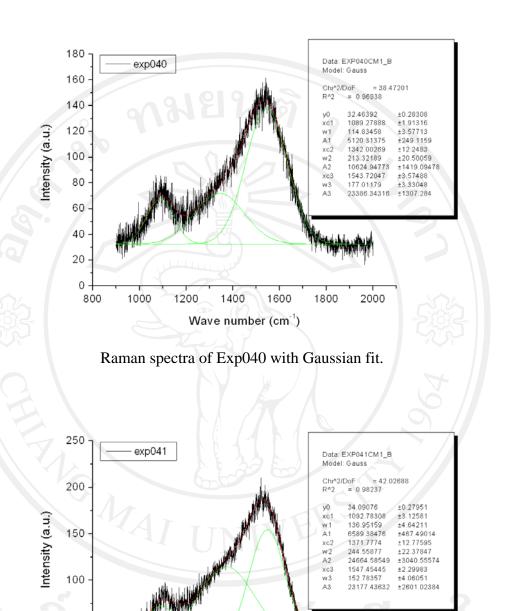






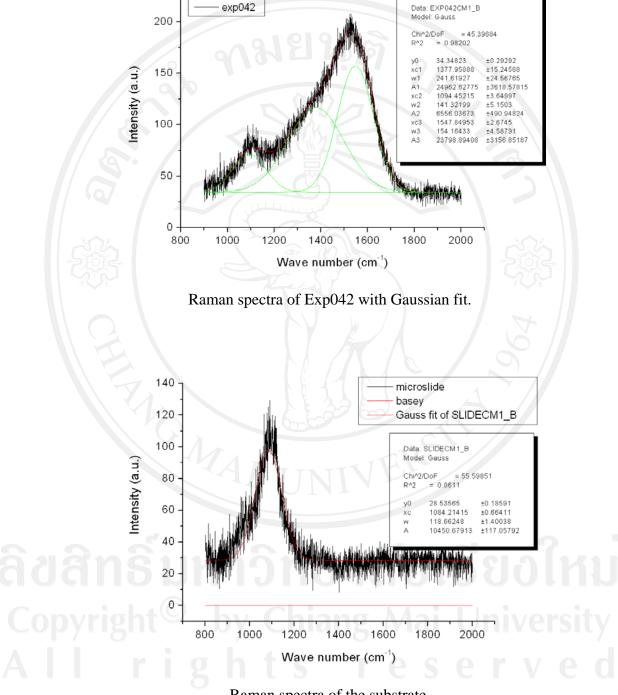






Raman spectra of Exp041 with Gaussian fit.

Wave number (cm⁻¹)



Raman spectra of the substrate.

APPENDIX B

Bipolar Plates

Appendix B-1: Photo of contact angle.

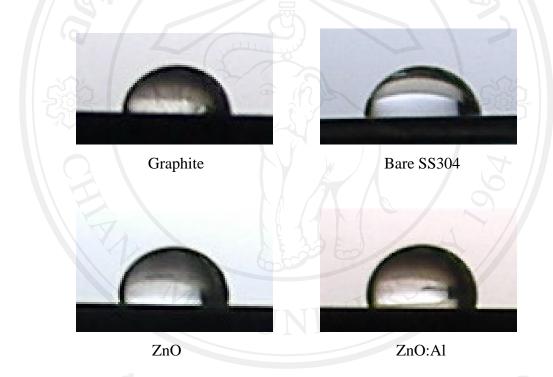




Photo of water droplet on specimen surface.

Appendix B-2: DOE bipolar plates technical targets.

Technical Barriers

(A) Durability

Improved corrosion resistance. Decrease weight and volume

(B) Cost

Lower material & production costs.

Increased power density due to decreased thickness

(C) Performance

Improved gas impermeability.

Improved electrical and thermal conductivity.

Technical Targets

The targets are listed in following table (Adrianowycz et al., 2008) is the DOE high temperature performance and low cost manufacturing targets for 2010 and beyond.

TABLE 1. DOE Technical Targets: Bipolar Plates (Table 3.4.14, Ref. 1)

Characteristic	Units	2010/2015	Project 2008 Status
Costa	\$/kW	5/3	TBD
Weight	kg/kW	<0.4	TBD
H₂ permeation flux	cm³ sec⁻¹ cm⁻² at 80°C, 3 atm (equivalent to <0.1 mA/cm²)	<2 x 10 ⁻⁶	TBD
Corrosion	μA/cm²	<1 ^b	<1 ^b
Electrical conductivity	S/cm	>100	>1,000
Resistivity	Ohm-cm	0.01	< 0.010
Flexural Strength	MPa	>25	>55
Flexibility	% deflection at mid- span	3 to 5	TBD

³ Based on 2002 dollars and costs projected to high volume production (500,000 stacks per year).

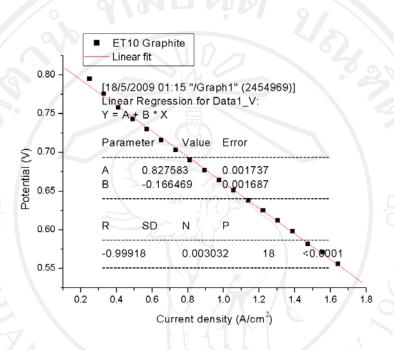
^b May have to be as low as 1 nA/cm if all corrosion product ions remain in ionomer for metal plates. Corrosion of flexible graphite plates is not an issue. ^c Includes contact resistance.

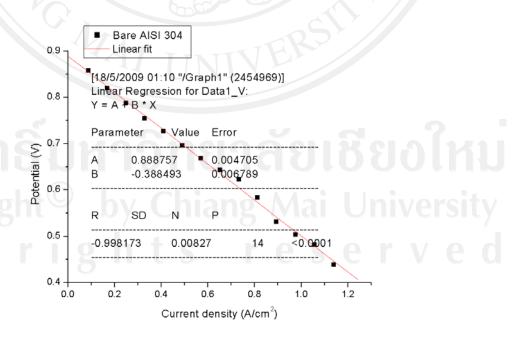
^dDevelopers have used ASTM C-651-91 using four point loading at room temperature.

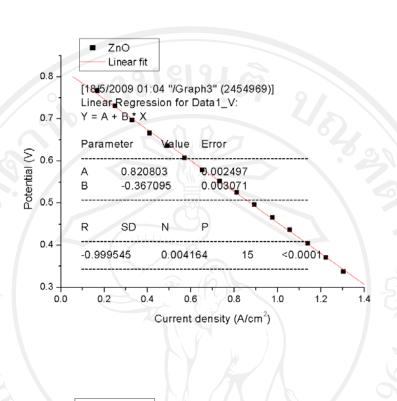
TBD - to be determined

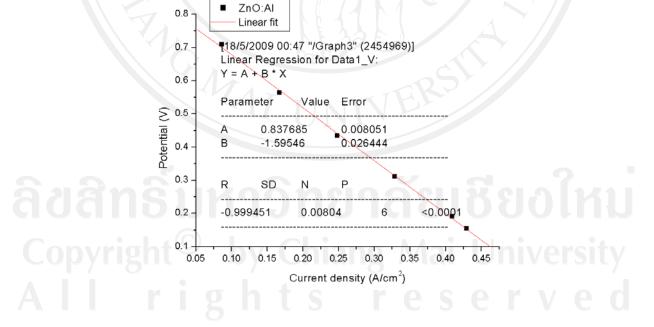
Appendix B-3: Linear fit of polarization curve.

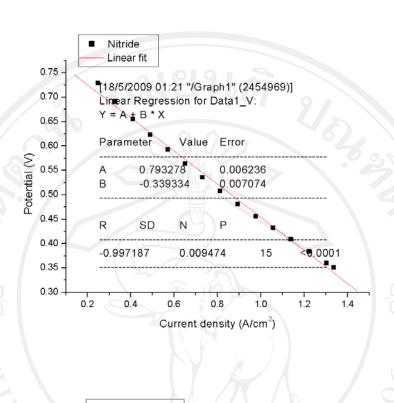
The linear fit of fuel cell polarization curve at center region (The cell potential drops linearly with current).

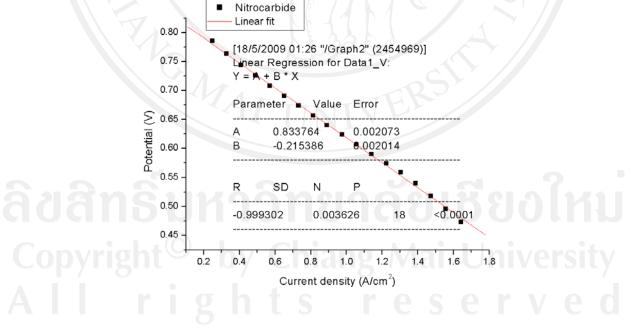












APPENDIX C

Publications and Presentation by Author

Publications:

- [1] **S. Sansongsiri**, A. Andres and B. Yotsombat, *Electrical properties of a-C: Mo films produced by dual-cathode filtered cathodic arc plasma deposition*, Diamond & Related Materials, Vol. 17, pp. 2080-2083, 2008.
- [2] N. Pasaja, **S. Sansongsiri**, S. Intarasiri, T. Vilaithong, A. Anders, *Mo-containing tetrahedral amorphous carbon deposited by dual filtered cathodic vacuum arc with selective pulsed bias voltage*, Nuclear Instruments and Methods in Physics Research Section B: Beam Interactions with Materials and Atoms, Vol. 256, pp. 867-870, 2007.
- [3] A. Anders, N. Pasaja and **S. Sansongsiri**, *Filtered cathodic arc deposition with ion-species-selective bias*, Review of scientific instruments, Vol. 78, 063901, 2007.
- [4] **S. Sansongsiri**, B. Yotsombat, Ramon E. Galindo, M. Rhodes, T. Vilaithong and A. Anders, *Pulsed d.c. glow discharge nitridation and carburization of stainless steel for polymer electrolyte membrane fuel cell bipolar plates*, submitted for publication in J. Power Sources.

Presentations:

- [1] **S. Sansongsiri**, B. Yotsombat and A. Anders, *Electrical properties of a-C:Mo films produced by dual-cathode filtered cathodic arc plasma deposition*, Poster presentation, CMU academic day 3, Chiang Mai University, Chiang Mai, Thailand, November 23-25, 2007.
- [2] **S. Sansongsiri**, R. Escobar Galindo, B. Yotsombat, T. Vilaithong and A. Anders, *Development of conductive and non-corrosive coating for PEM fuel cell bipolar plates*, Poster presentation, SIAM PHYSICS CONGRESS 2008, Nakhon Ratchasima, Thailand, March 20-22, 2008.
- [3] **S. Sansongsiri**, A. Andres and B. Yotsombat, *Electrical properties of a-C:Mo films produced by dual-cathode filtered cathodic arc plasma deposition*, Oral presentation, RGJ-Ph.D. Seminar LX, Chiang Mai, Thailand, March 28, 2008.
- [4] **S. Sansongsiri**, B. Yotsombat, R. Escobar Galindo, M. Rhodes, T. Vilaithong, and A. Anders, *Pulsed d.c. glow discharge nitridation and carburization of stainless steel for polymer electrolyte membrane fuel cell bipolar plates*, Poster presentation, Fuel Cell Science & Technology 2008, Copenhagen, Denmark, October 8-9, 2008.
- [5] **S. Sansongsiri**, B. Yotsombat, T. Vilaithong and A. Anders, *ZnO and ZnO:Al coated stainless steel bipolar plates for polymer electrolyte membrane fuel cell*, Poster presentation, CMU academic day 4, Chiang Mai University, Chiang Mai, Thailand, December 19-20, 2008.

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