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LIST OF ABBREVIATIONS

IC	Internal conversion
ISC	Intersystem crossing
BODIPY	Boron-dipyrromethene
ESIPT	Excited-state intramolecular proton transfer
HBI	2-(2'-Hydroxyphenyl)benzimidazole
HBO	2-(2'-Hydroxyphenyl)benzoxazole
НВТ	2-(2'-Hydroxyphenyl)benzothiazole
nm	Nanometer
HF G	Hartree-Fock approximation
СРСМ	The nonequilibrium polarizable continuum model calculations
DFT	Density functional theory
B3LYP	Becke, 3-parameter, Lee-Yang-Parr exchange-correlation
	functional
PBE0	Perdew, Burke, and Ernzerhof exchange-correlation functional
CAM-B3LYP	Handy and coworkers' long-range corrected version of B3LYP
	using the Coulomb-attenuating method
ωB97XD	Functional from Head-Gordon and coworkers including empirical
ລິບສາ	dispersion and a stable
LC-BLYP	The long correction of Hirao and coworkers
TD-DFT	Time-dependent density functional theory
НОМО	Highest occupied molecular orbital
LUMO	Lowest unoccupied molecular orbital
kcal·mol ⁻¹	Kilocalorie per mole
eV	Electron volts
ICT	Intramolecular charge transfer
PECs	Potential Energy Curves

- APBT 2-(2'-Aminophenyl)benzothiazole
- AzPBT 2-(2'-Azidophenyl)benzothiazole



LIST OF SYMBOLS

S	Excited singlet state
Т	Excited triplet state
υ	Frequency
\mathbf{S}_0	Ground state
S_1	First excited singlet state
S_2	Second excited singlet state
T1	First excited triplet state
0	Degree
R	Substituent
Ψ	Wavefunction
Ĥ	Hamiltonian operator
E	Total energy
Ψ	Wavefunction associated with solving the electron part
X	Wavefunction associated with nuclear motion
Т	Kinetic energy
V	Potential energy
$T_N(R)$	Kinetic energy of the nuclei
$T_e(r)$	Kinetic energy of the electrons
$V_{eN}(r,R)$	Electron-nuclei attractive Coulomb potential
$V_{NN}(R)$	Nuclear-nuclear repulsive Coulomb potential and
$V_{ee}(r)$	Electron-electron repulsive Coulomb potential
Ν	Many electron
$\psi_{i}(N)$	Spin-obital
\hat{f}	Fock operator
\mathcal{E}_i	Spin-obital energy
${\widehat H}^{(0)}$	Unperturbed Hamiltonian

∞	Infinity
ρ	Electron probability density
$\rho(r)$	Entire electron density of the molecule
E[ho]	Electronic energy as a function of electron density
$E_T[\rho]$	Kinetic energy as a function of electron density
$E_V[\rho]$	Potential energy as a function of electron density
$E_U[ho]$	External energy as a function of electron density
$E_{XC}[\rho]$	Exchange-correlation energy as a function of electron density
V _{XC}	Exchange-correlation potential
δρ	Integral of electron density
$E_{XC}^{Hybrid}[\rho]$	Hybrid functional energy of a mixture of Hartree-Fock exchange
	with DFT exchange-correlation
C	Constants determined by Becke
$\rho(r,t)$	Time-dependent potentials and electron densities
s	s-orbital
p	<i>p</i> -orbital
d	<i>d</i> -orbital
π	Pi AI UNIVER
$\pi\pi*$ and $\pi-\pi*$	Pi to pi* transition (proton transfer)
nπ* and n-π*	n to Pi* transition (intramolecular charge transfer)
Å Copvr	Angstrom
ΔΕ	Relative energy
%	Percent
f	Oscillator strength
Δλ	Deviation between the calculated and experimental wavelength

ข้อความแห่งการริเริ่ม

1) การออกแบบโมเลกุลสี่ข้อมฟลูออเรสเซนต์ตัวใหม่โดยวิธีการทางเคมีคอมพิวเตอร์ซึ่งเป็นมิตรต่อ สิ่งแวคล้อม ปราศจากสารเคมีที่เป็นพิษ ประหยัดเวลาและเป็นแนวทางสำคัญ ในกระบวนการ สังเคราะห์แก่นักเคมี

 การศึกษาสมบัติแสงเชิงฟิสิกส์และแสงเชิงเคมีของสีย้อมฟลูออเรสเซนต์ โดยการเติมเฮเทอ โร อะตอมหรือหมูแทนที่ เพื่อทำให้สเปกตรัมการคายแสงเรคชิฟท์ และทำให้ โมเลกุลมีส โตรกชิพมาก ขึ้นซึ่งเหมาะสมแก่การประยุกต์ใช้สีย้อมเหล่านี้ในเซลล์สิ่งมีชีวิต

AND MAI

STATEMENTS OF ORIGINALITY

1. Design the new molecular structure of fluorescent dyes by using computer chemistry, which environmental friendly, save time and this research are the guidance for chemists.

2. Photophysical and photochemical studies of fluorescent dyes. Theoretical studies of heteroatom and substituent effect resulting in the red-shifted emission spectra and large Stokes' shift, which can be applied for biological system.

