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LIST OF ABBREVIATIONS

ESPT Excited state proton transfer

ESIntraPT Excited state intramolecular proton transfer ESInterPT Excited state intermolecular proton transfer

PT Proton transfer

HBO 2-(2'-hydroxyphenyl)-benzoxazole
 HBT 2-(2'-hydroxyphenyl)-benzothiazole
 HBI 2-(2'-hydroxyphenyl)-benzimidazole

HBQ 10-hydroxy-benzo[h]quinolone

Tinuvin-P 2-(2'-hydroxy-50-methylphenyl- benzotriazole

3HC 3-hydroxychromone

3HF 3-hydroxyflavoes

E Enol form

E* Excited enol form

K Keto form

K* Excited keto form

nm Nanometer

s Second

fs Femtosecond

PMMA Poly methyl methacrylate

MTHF 2-methyltetrahydrofuran

MP 3-methylpentane

MCH Methylcyclohexane

MeOH Methanol

DFT Density functional theory

TD-DFT Time-dependent density functional theory

B3LYP Beck's three parameter Lee-Young-Parr

TZVP Triple zeta valence plus polarization

BOA Born-Oppenheimer approximation

HF Hatree-Fock

CI Configuration Interaction

MPn Perturbation theory

CC Coupled-cluster methode

QMC Quantum Montecarlo

SCF Self-consistent field

STOs Slater-type orbitals

GTOs Gaussian-type orbitals

DZ Double-zeta

TZ Triple-zeta

TDSE Time-dependent Schrödinger equation

PECs Potential energy curves

NH₃ Ammonia

CH₃OH Methanol

H₂O Water

MOs Molecular orbitals

HOMO Highest occupied molecular orbital

LUMO Lowest unoccupied molecular orbital

by Chiang Mai University

LIST OF SYMBOLS

S₀ Ground state

S₁ First excited state

S₂ Second excited state

T₁ Triplet state

-OH Hydroxy group

-C=O Carbonyl group

Ĥ Hamiltonian operator

Ψ Wavefunction

 T_n Kinetic energy of nuclei

 T_e Kinetic energy of electron

V_{en} Electron-nuclear interaction Coulomb potential

 V_{nn} Repulsive Coulomb potential of nuclear-nuclear

Vee Repulsive Coulomb potential of electron-electron

h Plank constatnt

M Mass of nucleus

m Mass of the electron

R_{AB} Distance between nuclei A and B

 r_{ij} Distance between electrons i and j

r_{iA} Distance between electron *i* and nucleus A

 ε_0 Permittivity of free space

Z Nuclei charge

 χ_i Spin orbitals

α/β Electron spin up/down

x_i Coordinates of the electron i

 v^{eff} Effective potential of each electron

 $\rho(\mathbf{r})$ Total electron density of molecule

 $E[\rho]$ Electronic energy as a function of electronic density E_T Kinetic energy as a function of electronic density

E_V Interaction energy between electron and nuclei

E_J Coulomb energy

Exchange-correlation energy

 E_{XC}^{hyb} Hybrid function energy of a mixture of Hartree-Fock exchange with

DFT exchange-correlation

 $\rho(r,t)$ Time-dependent potentials and electron densities

 P_w A Wigner distribution

Pⁱ Momentum associated with the normal coordinate

Oⁱ Normal coordinate

 ξ_{HO}^n The harmonic oscillator wavefunction

s s-orbital

p p-orbital

d d-orbital

 π Pi

 π - π * Pi to Pi* transition (proton transfer)

Å Angstrom

% Per cent

f Oscillator strength